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Plant growth regulators (PGRs): Providing modern solutions to conventional agricultural problems

Farwa Nadeem*

Nano and Biomaterials Lab, Department of Chemistry, University of Agriculture, Faisalabad-38040-Pakistan.

Abstract

Rapid increase in global population is a constant threat to the food security all over the world. The only way to cope with the ever increasing demands of food crops and aromatic medicinal plants is to modernize the conventional agricultural practices. The use of cost efficient, effective and environmentally benign plant growth regulators (PGRs) appears to be the most appealing option for the modulation of plant growth. The plant growth regulators (PGRs) work by altering the physiological traits and enhancing the plant's response towards biotic and abiotic stresses. This review article basically discusses the role of plant growth regulators (PGRs) in enhancing the yield of essential oil and major secondary metabolites of aromatic medicinal plants. The data is summarized for some selected plant growth promoters (PGPs) (such as (a) sodium nitropheno-late (b) 28-homobrassinolide (c) 1-nephthylacetamide (d) oxalic acid and (e) hymexazol) and plant growth inhibiters (PGIs) (such as (a) chlormequat chloride (b) maleic hydrazide (c) daminozide (d) ethephon and (e) tetcyclacis). However, some major problems in the use of plant growth regulators (PGRs) include photodegradation, thermal degradation, cytotoxic effects and excessive leaching of applied chemicals that can be overcome by using nanotechnology. In the last section of review article, some advanced spectroscopic techniques (such as Fourier transform infrared spectroscopy (FTIR), Lased induced breakdown spectroscopy (LIBS), Atomic absorption spectrometry (AAS), Scanning electron microscopy (SEM), Thermal gravimetric analysis (TGA) and X-ray diffraction (XRD)) are mentioned to study the complete chemical composition and surface properties of plant growth regulators (PGRs).

Keywords: Plant growth regulators, sodium nitrophenolate, chlormequat chloride, LIBS, TGA

Full length article *Corresponding Author, e-mail: <u>farwa668@gmail.com</u>

1. Introduction

The conventional agricultural practices face many challenges in fulfilling the rapidly increasing food demands of growing global population. Therefore, significant improvements in old agricultural practices are needed to ensure the high quality nutrition for living beings [1]. The adequate food production and proper cultivation of aromatic medicinal plants can only be made possible by increasing the percentage of arable land area. The use of environmentally benign plant growth regulators (PGRs) is one of the most appealing options in order to modulate the plant growth and development. The plant growth regulators (PGRs) are the chemical substances having specific formulations, capable of modifying the physiological traits and promoting the improved stress responses of plants towards external stimuli. These plant growth regulators (PGRs) can improve the nutritional profile and maximize the production of crops, by overcoming the genetic problems and biotic and abiotic stresses of plants [2]. Some conventionally used plant hormones or plant growth Nadeem, 2022

regulators (PGRs) include auxins, cytokinins, paclobutrazol, mepiquat chloride, brassinolide and many structural analogues. The excessive utilization of plant growth regulators (PGRs) in agriculture came to the limelight within past two decades as their benefits were understood by the local growers and poor farmers. Unfortunately, the growth of the PGR market may be constrained by a lack of innovation at a time when an increase in demand for new products will require steady innovation and discovery of novel, costcompetitive, specific, and effective PGRs [3]. Therefore, the present review article discusses the biosynthetic pathways involving the production of essential oils and menthol contents in aromatic medicinal plants with special emphasis on Mentha arvensis L. The second major part summarizes the data on some selected plant growth promoters (PGPs) (such as sodium nitrophenolate, 28-homobrassinolide, 1nephthylacetamide, oxalic acid and hymexazol) and plant growth inhibiters (PGIs) (such as chlormequat chloride, maleic hydrazide, daminozide, ethephon and tetcyclacis). The third part of review article includes the problems associated with the use of plant growth regulators including photodegradation, thermal degradation and cytotoxic effects and leaching potentials. Some of the advanced characterization techniques are also mentioned at the end of review article.

2. Biosynthesis of essential oils in plants

In almost all the aromatic plants of kingdom plantae, secondary metabolites are produced for the purpose of (i) formation of ozone of troposphere (ii) formation of fine aerosol particles (iii) quenching of ozone (iv) thermotolerance of plants (v) attracting the insects for the purpose of pollination and dispersion of seeds (vi) plant to plant signaling and (vii) protection against the foreign invaders and pathogens. Essential oils are one of the most important phytochemicals and secondary metabolites of plants, having natural aroma and high volatility under all conditions of temperature and pressure. These volatile organic compounds usually originate from three types of chemical compounds including the (a) isoprenoids (b) fatty acid derivatives and (c) phenolic compounds. Essential oils are highly complex in their chemical composition due to the presence of diverse class of compounds like phenylpropanoids, sesquiterpenoids, and monoterpenoids. In all aromatic plants, essential oils are synthesized by two natural and highly complex biochemical pathways, involving the chains of enzymatic reactions. For the biosynthesis of essential oils in plants, isopentenyl diphosphate (IPP) and dimethylallyl diphosphate (DMAPP) are the two major and universal precursors. These compounds are produced by two possible pathways (i) plastidic and enzymatic 1-deoxy-D-xylolose-5-phosphate (DXP) pathway, commonly termed as the 2-C-methylerythritol-4-phosphate (MEP) pathway and (ii) the cytosolic enzymatic mevalonic acid (MVA) pathway. In different parts of cell, prenyl diphosphate synthases immediately condenses into the dimethylallyl diphosphate (DMAPP) and isopentenyl diphosphate (IPP), for the further conversion into prenyl diphosphates. The prenyl diphosphate is a basic substrate for the synthesis of fernesyl diphosphate (FPP) and geranyl diphosphate (GPP). All the major constituents of the essential oils are final products of terpenoids, and are ultimately formed by the complex group of enzymes also known as terpene synthases (TPS) [4]. The detailed mechanism of the synthesis of essential oils in plants is shown in (Fig.1).

3. Biosynthesis of menthol in Mentha arvensis L.

The epidermal tissues of the plants contain two different types of trichomes named as (a) simple trichomes (STs) and (b) glandular secreting trichomes (GSTs). The STs are also known as "non-glandular trichomes" and they are directly associated with the gene expressions, involved in the biosynthesis of glucosinolate, flavonoid and anthocyanin. The glandular secreting trichomes (GSTs) significantly influence the water retention, pollinator attraction, pest resistance and pathogen defence of the plants, depending upon the phytochemical secretions. These GSTs plays an important role in synthesis of natural products in aromatic plants. In case of *Mentha arvensis* L., special "peltate glandular trichomes" are found on the aerial parts of plant, and directly involved in the secretions of essential oils. In this plant, eight radially distributed secretory cells are located on the young stalk and also embedded in the basal cells. These cells produce the essential oils, involving the number of complex biochemical reactions, as per the mechanism shown in (Fig.2) [5].

4. Plant growth regulators (PGRs)

A diverse group of naturally occuring or synthetically produced larger complex organic compounds, having significant influence on the growth parameters and secretions of secondary metabolites, specifically essential oil contents of ornamental and medicinal aromatic plants are termed as plant growth regulators (PGRs) [6]. The plant growth regulators (PGRs) are considered to be highly proficient elicitors for the stimulation of production of secondary metabolites in plants [7]. Sometimes, plant growth regulators (PGRs) are confused with natural plant hormones, but there are certain differences in both these chemicals. The plant growth regulators (PGRs) are specific agrochemicals that are mostly produced synthetically and applied to the plants to enhance the required phytochemicals and secondary metabolites through modification of biochemical pathways, while the natural plant hormones directly influence the different physiological processes of plants [8]. The production of secondary metabolites in plants is believed to be stimulated under stressful environmental conditions, and thus application of plant growth regulators (PGRs) is the most probable option to enhance the essential oil contents, more specifically menthol contents of Mentha arvensis L. [9].

4.1. Plant growth promoters (PGPs)

Plant growth promoters (PGPs) are mostly synthesized and transported through the vascular system of plants, but any change in external environment can lead to the modification of cellular response. The living cells of plants try to balance the externally applied plant growth promoters (PGPs), either through up-regulation or down-regulation, depending upon the extent of change in the surrounding Therefore, targeted delivery environment. of the nanostructured plant growth promoters (PGPs) can ensure the expected developmental responses [10], through increased efficiency of plant growth promoters (PGPs) and highly controlled delivery of nutrients with the minimum usage [11]. According to the European Biostimulant Industry Council

(EBIC), "biostimulants" are the chemical substances that stimulate the natural processes of plants to (i) increase the quality of crops (ii) tolerate the abiotic stresses and (iii) enhance the nutrient uptake efficiency [12]. The effects of sodium nitrophenolate, 28-homobrassinolide, 1nephthylacetamide, oxalic acid and hymexazol on the essential oil contents and growth characteristics of different aromatic and non-aromatic plants are compiled in table 1.

4.2. Plant growth inhibitors (PGIs)

Salt stress, acute water shortage, high temperature, saline conditions, insect attacks and harsh environmental conditions can significantly increase the release of secondary metabolites in plants [13]. Therefore, application of growth retardants or plant growth inhibitors (PGIs) not only reduces shoot elongation, but also improves the chemical composition of essential oils in aromatic medicinal plants. The plant growth inhibitors (PGIs) can be classified into two major groups (i) chemicals inhibiting the biosynthesis of gibberellic acid (GA) and (ii) chemicals releasing the ethylene based compounds. Until now, four different types of plant growth inhibitors (PGIs) have been used including (a) 16,17-Dihydro-GA₅ and all the structurally related compounds that acts like gibberellic acid (GA) using dioxygenases (b) structurally similar compounds of 2-oxoglutaric acid that generally involves in the catalysis of series of biochemical reactions, resulting in formation of gibberellic acid (GA) (c) nitrogen containing heterocyclic compounds that are capable of blocking cytochrome P₄₅₀-dependent monooxygenases, involved in the oxidation of ent-kaurene into ent-kaurenoic acid and (d) onium compounds that block the ent-kaurene synthase and cyclases copalyl-diphosphate synthase, directly involved in metabolism of gibberellic acid (GA) [14]. The effects of chlormequat chloride, maleic hydrazide, daminozide, ethephon and tetcyclacis on the essential oil contents and growth characteristics of different aromatic and non-aromatic plants are compiled in table 2.

5. Importance of nanotechnology in agriculture

The use of nanotechnology and nano-scale formulations can significantly alter the global agricultural canvas, by providing effective solutions, for a number of agriculture-related-problems. Nanoparticles act as a bridge in between atomic/molecular structures and their bulk counterparts, and thus are known to have great scientific interest. Over the past few decades, huge amount of work has been carried out on the use of nanotechnology in agriculture, emphasizing its number of applications in this sector [11-15-17]. The use of nano-carriers for the application of plant growth regulators (PGRs) can ensure the slow delivery and sustained release of bioactive components, thereby avoiding their supra-optimal levels. In addition to this, nanoformulations can improve the solubilizing potentials of plant growth regulators (PGRs) by protecting them from photodegradation, thermal degradation, and biodegradation [18-21].

Nanomaterial engineering is considered to be a major cutting-edge-track of the recent scientific researches, supporting the development of advanced agricultural techniques, through providing a larger specific surface area, essentially required for the sustainable development of agricultural system [11-22-23]. Nanotechnology not only decreases the uncertainty, but also helps in the management of agricultural production, as an effective alternative to the conventionally used agricultural technologies. The innovations in agro-nanotechnology offer the short term solutions for a number of agricultural problems, being faced in the modern system of agriculture [11]. Nanoencapsulations have full potential to increase the bioavailability, improve the sustained delivery of applied chemicals and enable the targeted supply of bioactive components, as compared to micro-encapsulations [24-25].

Therefore significant improvements and continuous developments are ultimately required for the production of nanoparticles, having potential applications in the modern environmental and agricultural systems [26]. In case of conventional agricultural practices, the agrochemicals are mostly applied to the crops using broadcasting and foliar sprays. These types of systems enable the small amount of agrochemicals to reach at the actual target, resulting in the supply of doses much lower than the effective concentrations of agrochemicals required for the proper growth and development of plants. Most of the agricultural loses are mainly due to the microbial degradations, hydrolytic reactions, photolysis and leaching of applied agrochemicals [27-28]. In order to ensure the environmental friendly agricultural practices, recent developments are required in the synthesis of nanotechnology based herbicides, pesticides, fungicides, fertilizers, and plant growth regulators (PGRs) [22].

The use of nanoparticles in the modern system of agriculture is highly advantageous due to the effective delivery of agrochemicals at the targeted location, mainly because of the larger surface area, high mass transfer rate and easy attachment of applied chemicals [29]. One of the major difficulties in the use of plant growth regulators (PGRs) is rapid degradation on exposure to the number of environmental factors such as high temperature and intense light, resulting in the loss of natural activity. Another major advantage of the use of nanotechnology in the agriculture is to provide the nanoscale materials, capable of enhancing the bioactivity and stability of bioactive agents, along with the significant reduction in harmful environmental consequences. Therefore, encapsulation of bioactive agents can improve the physiochemical characteristics, and avoid

the excessive leaching. Nanotechnology also ensures the maximum bioavailability and minimum wastage of agrochemicals, making it safer for the environment [19].

The controlled release of agrochemicals is a "permeation-regulated-transfer" of bioactive ingredients from the source reservoir to the targeted surface, for maintenance of predetermined concentrations of agrochemicals, for a specific period of time. Some major objectives of the use of nanotechnology in agriculture are (i) high reliability of entire process (ii) improved safety of environment (iii) prolonged efficacy (iv) negligible side effects, and (v) very effective treatments [30-31]. Over the past few decades, much literature has been published regarding the use of plant growth regulators (PGRs) and their nano-encapsulations on different aromatic and non-aromatic plants, but all these methods and processes are quite prolonged or time consuming, inexpensive or cost inefficient, energy consuming, and uses high-tech instruments, making it unfit for the local growers and poor farmers. Therefore, most of these processes are only restricted to the scientific researches and peer reviewed articles, and cannot be commercialized properly. Some nano-structured plant growth regulators (PGRs) and their methods of preparation with complete reaction conditions are mentioned in table 3.

6. Factors affecting the plant growth regulators (PGRs)

6.1. Photodegradation

Photodegradation of plant growth regulators (PGRs) and other agrochemicals is one of the major challenges faced by global agriculture. The amount of solar radiations required for the photolysis of plant growth regulators (PGRs) mainly depends on the ultraviolet absorption profile of plant growth regulators (PGRs), emission spectrum of sunlight, and chemical composition of surrounding media. The minimum amount of energy required to break the chemical bonds of plant growth regulators (PGRs) generally ranges from 70 kcal/mol to 120 kcal/mol, corresponding to the wavelength of sunlight in the range of 250 to 400 nm. Therefore, the solar radiations reaching towards the surface of ground are very important in determining the photodegradation profiles of plant growth regulators (PGRs) and pesticides. The intensity of solar radiations decreases up to 10% by passing through the atmosphere and reaching towards the troposphere. At the wavelength ranging from <290 to >295 nm, most of the solar radiations are absorbed by the layer of ozone, prior to reaching the surface of ground. Therefore, all the solar radiations near the ground surface exhibit the wavelengths ranging from 440 to 460 nm, and its ability to cause the photodegradation in plant growth regulators (PGRs) becomes only 5 to 6% of the total degradation intensity [32].

The absorption of sunlight by any material is known to have number of photo-physical pathways, involving the complex system of interactions. According to Franck-Condon principle, solar radiations passing through the plant growth regulators (PGRs) causes molecular excitations, due to the interactions in between sunlight and electric field of plant growth regulators (PGRs), at the timescale of femtoseconds, without showing any change in the molecular geometry of the agrochemical. As per the Stark-Einstein rule, single photon can excite only one molecule of the agrochemical, with the probability of excitation in the lowest excited state (S₁), which further involves number of photophysical processes. Usually, it has been observed that the molecules of plant growth regulators (PGRs) exhibit ultraviolet-visible spectrum at wavelength >290 nm, due to the presence of aromatic substituted moieties. In some cases, plant growth regulators (PGRs) are known to contain the unsaturated bonds and lone pairs of electrons of carbamoyl or carbonyl groups, thus showing the $n \rightarrow \pi^*$ or $\pi \rightarrow \pi^*$ transitions upon irradiation [33].

According to the recent photo-physical studies, there are only three possible pathways for the excitation of a molecule from lower energy level to higher energy states (Fig.3) (i) non-radiative internal conversion (IC) system (ii) emissions due to fluorescence and (iii) intersystem crossing (ISC) to reach up to the excited triplet state (T_1) . In first pathway, all excited molecules are relaxed from the higher vibrational levels to the S_1 state within 10^{12} sec⁻¹, followed by the decay towards the lower electronic energy states within 10^6 to 10^{12} sec⁻¹. In the second pathway, all the excited molecules undergo radiative deactivation through fluorescence, and fluorescence spectra is found similar to the absorption spectrum caused by the Franck-Condon principle, but with the shift towards the red colour. The duration of florescence ranges from few nano-seconds to few microseconds, due to the transitions in between similar energy states. In third pathway, spin-forbidden process $(S_1 \rightarrow T_1)$ is followed with the help of (a) emission of phosphorescence and (b) radiationless deactivation. In $S_1 \rightarrow T_1$ process, the duration of phosphorescence generally ranges from few milliseconds to $10^2 \sec [34-35]$.

In the higher energy state, all the excited molecules can cause different types of chemical reactions, unless their total energy contents are lost by the emission of light and release of heat. There are two possible forms of photochemical reactions such as (a) "direct photolysis" and (b) "indirect photolysis". In the case of direct photolysis, photochemical reactions takes place by the absorption of light energy, while in case of indirect photolysis, the chemical reactions of ground state molecules are preceded by the excitation with other excited molecules and photochemical species. The direct photolysis is also termed as "quenching" or "photosensitization" and indirect photolysis is the "photo induced reaction with reactive oxygen species" [32]. Some recent photodegradation studies of plant growth regulators (PGRs) and other agrochemicals are compiled in table 4.

6.2. Thermaldegradation

Thermaldegradation of plant growth regulators (PGRs) and other plant nutrients significantly limit the agricultural output and overall productivity of aromatic plants. Therefore, thermal stability of agrochemicals plays an important role in determining the shelf-life, longterm utilization and efficiency of plant growth regulators (PGRs). Thermal analysis (TA) is used to study the kinetic properties, thermo-physical characteristics, total char contents and decomposition temperature of agricultural chemicals, by measuring the change in mass with respect to change in temperature. The temperature range usually varies from -150 to 1600°C depending upon the nature of sample and the type of information required. For the preparation of novel metal based nano structured plant growth regulators (PGRs), all temperature ranges including the degradation temperature must be known, for heating up to specific level. Many techniques have conventionally been used in thermal analysis (TA) such as (i) pressurized thermogravimetric analysis (PTGA) (ii) dilatometry (DIL) (iii) laser flash analysis (LFA) (iv) evolved gas analysis (EGA) (v) dielectric thermal analysis (DEA) (vi) differential thermal analysis (DTA) (vii) dynamic mechanical analysis (DMA) (viii) thermo mechanical analysis (TMA) (ix) differential scanning calorimetry (DSC) and (x) thermogravimetric analysis (TGA) [36].

All these techniques are based on variations in temperature with respect to change in (a) PTGA-change in mass as a function of pressure (b) DIL-change in volume (c) LFA-change in thermal conductivity and thermal diffusivity (d) EGA-change in chemical composition of final gaseous products (e) DEA-change in loss factor and dielectric permittivity (f) DTA-difference in temperature (g) DMAchange in damping and mechanical stiffness of analyte of interest (h) TMA-change in dimensions and deformations (i) DSC-difference of heat and (j) TGA-change in mass. Among all these methods, TGA is the most sensitive and highly recommended method for the agricultural chemicals, to study the change in structure and composition of final product. It involves the heating of plant growth regulators (PGRs) till its degradation point, with gradual increase in temperature. Similarly, DSC is another important thermodynamic tool for direct estimations of change in heat energy with the regular increase or decrease in temperature. DSC is capable of measuring the heat capacity, specific heat, energy of precipitation, oxidation induction time, crystalline phase, transition temperature, energy of reaction, melting temperature, glass transition temperature, latent heat of melting and heat of fusion [36]. Some recent data on the Nadeem, 2022

thermal degradation of plant growth regulators (PGRs) and other agrochemicals is listed in table 5.

6.3. Leaching potential

Nitrogen, phosphorous and potassium are the critical elements essentially required for the increased forage production, improved fertility, and higher crop yield. However, the build-up of surplus agricultural nutrients for increased forage demands can lead to the leaching of nutrients, thereby causing the economic losses, wastage of nutrients and significant harm to environment. This type of nutrient leaching can cause the soil degradation, water contamination, air pollution, and eutrophication involving the loss of nutrients through the surface water. The air pollution is caused by the emissions of greenhouse gases, more specifically ammonia and carbon dioxide. Some additional environmental problems include all the concerns regarding the sustainability of diminishing phosphorous resources from soil [37-38].

Similarly, plant growth regulators (PGRs) plays an important role in development of effective responses in plant as a result of abiotic stresses [39]. The plant growth regulators (PGRs), pesticides, herbicides, fungicides and various other agrochemicals are found to persist in the environment for relatively long period of time, with the capacity of bioaccumulation through food web. These agrochemicals, specifically organochlorine pesticides can directly affect the human health and surrounding environment. The extent of persistence of agrochemicals and their harmful consequences mainly depends on the (i) toxicity of compound (ii) type of formulation (iii) amount of chemical applied (iv) their mode of application and (v) mobility/incorporation in environment. Some agrochemicals also acts as an "endocrine disruptors" that alter the functions of endocrine system [40].

Leaching, degradation and adsorption are some of the key processes, having significant influences on the fate of agrochemicals in soil. Some important processes like (i) plant uptake (ii) assimilation by microorganisms (iii) water runoff (iv) soil erosion (v) volatilization (vi) diffusion and (g) leaching are mainly responsible for the movements of agrochemicals in environment. Leaching can be defined as downward displacement of agrochemicals in soil by passing through the unsaturated zone and finally reaching towards groundwater, thereby causing contamination in the natural water resources. Irrigation water and excessive rain is mainly responsible for leaching and its rate is highest for the persistent compounds, being used in the area of high precipitation and low temperature. The extent of leaching depends on the nature of agrochemical, type of soil, rate of volatilization, crop-root uptake, method of application and type of organisms inhibiting the soil [38]. Leaching of some important agrochemicals is compiled in table 6.

6.4. Cytotoxic effects

The 3-[4,5-dimethylthiazol-2-yl]-2,5diphenyltetrazolium bromide (MTT) assay is a colorimetric method for the measurement of metabolic activity of living cells. It is mainly based on the natural ability of "nicotinamide-adenine-dinucleotide-phosphate" (NADPH) dependent oxidoreductase enzymes of the cells to completely reduce the soluble MTT reagent (light yellow) into insoluble formazan crystals (dark purple) (Fig.4). Therefore, MTT assay measures the viability of living cells as it converts the "water soluble tetrazolium compound" into "water insoluble formazan crystals" with the help of mitochondrial dehydrogenases. In this cytotoxic assay, sodium dodecyl sulfate, acidified ethanol and dimethyl sulfoxide are used to dissolve the insoluble formazan crystals and to get the coloured solution [41].

The obtained coloured solution shows maximum absorption at the wavelength ranging from 500 to 600 nm, by using the UV-Vis spectrophotometer. The MTT reagent is taken up by the living cells through endocytosis and immediately reduced into formazan crystals, by using the endosomal, lysosomal and mitochondrial enzymes, inside the cellular compartments. However, living cells can dramatically be damaged by complete metabolism of MTT reagent, as it can activate the "apoptosis-related-factors" including the caspase-3 and caspase-8, resulting in the leakage of cellular components [42]. The cytotoxic effects of various *Mentha* species on different cell lines are compiled in table 7.

7. Advanced characterization techniques

The physical properties and chemical reactivity of nanomaterials is quite different from their bulk counterparts mainly due to high "surface-to-volume ratio" and "peculiar morphological characteristics". Preparation of nanomaterials is still under the developmental stages, and so the analytical methods, used for the complete characterization of nanoparticles (NPs). The size and shape are the two major parameters mostly considered for the improved reactivity of nanoparticles (NPs). Up till now; there are no standard protocols to study the chemical composition and crystal structure of nanoparticles (NPs). Thus, credible and robust methods of measurements for the nanoparticles (NPs) can enhance the commercial scale utilization of nanomaterials. Some major challenges faced by the nanotechnology includes (i) interdisciplinary nature of nanotechnology (ii) lack of proper reference materials (iii) improper calibration of analytical tools (iv) difficult sample preparation and (v) complex interpretation of obtained spectra. Therefore, manufacturing of nanoparticles (NPs) require highly reliable quantification methods for the characterization of nanoparticles (NPs) [43]. Some of the major characterization Nadeem, 2022

techniques for the nano structured plant growth regulators (PGRs) are discussed in detail in the table 8.

7.1. Fourier transform infrared spectroscopy (FTIR)

Fourier transform infrared spectroscopy (FTIR) is an effective method and important tool for the measurement of absorption of electromagnetic radiations, mainly in the mid infrared (IR) region of electromagnetic spectrum. This region of electromagnetic spectrum generally ranges from 4000 to 400 cm⁻¹. The absorptions in this region are based on changing dipole moment and vibrations of IR active molecules. In the IR spectra, positions of bands are mainly relevant to the nature and strength of bonds, and functional groups present in the analytical samples. This type of spectroscopy provides valuable information about the molecular structures and complex interactions in between different functional groups [43].

7.2. Laser induced breakdown spectroscopy (LIBS)

Laser induced breakdown spectroscopy (LIBS) is an emerging analytical technique, used for the *in-situ* elemental analysis of various organic, inorganic and organometallic compounds. This technique is based on the emission spectroscopy [44] and thus measures the spectral emissions of different elements, in the plasma generated by high energy laser. LIBS have been used for elemental analysis since past few decades due to the easy sample preparation, simultaneous multi-element detection, and rapid responses [45]. LIBS require the minimum pretreatment of samples, produces no unwanted byproducts, and generates the spectra based on excitations and de-excitations of species at specific wavelength [44].

7.3. Atomic absorption spectrometry (AAS)

Atomic absorption spectrometry (AAS) is another useful analytical technique used for the elemental analysis of metallic compounds. Generally, this technique can be employed with the help of three different atomization processes like chemical vapor generation atomic absorption spectrometry (CVG-AAS), electrothermal atomization atomic absorption spectrometry (ETAAS) and flame atomic absorption spectrometry (FAAS). The selection of appropriate method depends on the chemical nature of analytical sample and physical state of major chemical constituents. For the direct determination of analyte of interest in the solid and liquid samples, AAS requires some specific modifications in the basic instrumentations, to ensure the reliability of results and obtained informations [46].



Fig.1: Biosynthetic pathway of essential oil in plants [4]



Fig.2: Biosynthesis of secondary metabolites of Mentha arvensis L. [5]



Fig.3: Jablonski diagram showing different photo-physical pathways [33]



Fig.4: Biochemical conversion of MTT reagent into formazan crystals [49]

Sr. No	Plant species/Type of	Name of chemical applied	Effects on essential oil	Effects on other growth	References			
	study		contents/Miscellaneous effects	attributes/Additional properties				
			Sodium nitrophenolate					
			N ⁺ O Na H ₃ C					
1	Amaranths hypochondriacus L.	Sodium nitrophenolate in compound form	Significant decrease in concentration of cadmium from highly polluted land area due to the excessive grown plants	Improved enzymatic potentials of catalases in leaves and higher biomass yield and chlorophyll contents in grains	[50]			
2	Barley fodder	Atonik	Use of hydroponic solution for barley as a basic growth media, instead of soil	Hydroponic solution improved the dry matter, acid detergent fibre, neutral detergent fibre, crude protein, crude fat and crude fibre contents	[51]			
3	Mentha piperita	Atonik also known as Asahi SL	Significant increase in essential oil percentage	Marked increase in herbage yield of plant	[52]			
	1		28-Homobrassinolide	P				
	28-Homobrassinolide H_3C , H_3C ,							
1	Mentha arvensis L.	28-homobrassinolide	Significant increase in percentage of major secondary metabolites such as menthyl acetate, menthone and menthol in essential oil	Higher carotenoids, flavonoid and chlorophyll contents with significant increase in fresh and dry weight of	[53]			

Table 1: Effects of PGPs on essential oil contents and growth parameters of different plant species

				plants, lead area index and height of mint		
2	Mentha arvensis L.	28-homobrassinolide	Appreciable increase in percentage of menthyl acetate, l-menthone, isomenthone and menthol contents	Significant improvement in rate of photosynthesis, stomatal conductance and overall biomass yield	[54]	
3	Mentha arvensis L.	28-homobrassinolide	High percentage of isomenthone, menthyl acetate, menthol and l- menthone	Significant improvements in herbage yield and physiological characteristics	[55]	
I-Naphthylacetamide						
1	Photodegradation and photocatalytic potentials	2-(1-naphthyl) acetamide	Characterization of plant growth regulator by using liquid chromatography electrospray ionization tandem mass spectrometry	Plant growth regulator and potential pesticidal compound	[56]	
2	Studies involving the photo-stabilization	2-(1-naphthyl) acetamide	Two types of spectroscopic techniques (i) absorption spectroscopy and (ii) fluorescence spectroscopy	Encapsulation of plant growth regulator in the envelop of β- cyclodextrin for the photo- stabilization	[57]	
3	Photo-physical characterization of plant growth regulator	2-(1-naphthyl) acetamide	Time resolved spectroscopy and steady state spectroscopy	2-(1-naphthyl) acetamide is a synthetic plant growth regulator which mimic the functioning of natural indole auxin	[58]	
	<u> </u>		Oxalic acid H H O H	· I		

1	Tomato fruit	Oxalic acid	Significant improvements in gene	Marked improvements in defense	[59]
			oncological studies	metabolic stresses, phosphorylation	
			oneological studies	and oxidation reduction reactions	
				along with enhanced chilling tolerance	
2	Shoots of Bamboo Plant	Oxalic acid	Improved post-harvest management	Enhanced membrane integrity	[60]
-	(Phyllostachys		and quality of plant material with	reduced respiration retarded	[00]
	nrominens)		marked increase in chilling	lignification process decreased losses	
	prominens)		responses	of sugar contents, inhibited enzymatic	
			responses	browning and decreased disease	
				incidences	
3	Pomegranate fruit	Oxalic acid	Improved chilling tolerance and	Remarkable increase in total phenolic	[61]
-	8		better post-harvest quality of fruits	contents, catalase activity and	[*-]
			I I I I I I I I I I I I I I I I I I I	antioxidant potentials	
	•		Hymexazol		
			H ₃ C NH		
1	Increased pesticidal activities with negligible process losses	Hymexazol	Synthesis of hymexazol-graphene oxide-polydopamine based nano- composites	A known plant growth regulator and a potential antifungal compound	[62]
2	Preparation of host-guest- inclusion-complexes using hymexazol	Hymexazol	Significant growth promoting effects and pronounced antifungal activities	Mycelial growth of fungal strains is reduced by the exogenous application of hymexazol in complex form	[63]
3	Estimation of antifungal	Hymexazol	Synthesis of hymexazol-linked-	Hymexazol has been proved as a	[64]
	potentials of hymexazol-		chitosan derivatives	potential antifungal agent for number	
	linked-chitosan			of medicinal aromatic plants against	
	derivatives			<i>Gibberella zeae</i> CGMCC _{3.42} and	
				Rhizoctonia solani CGMCC _{3.28}	

Sr. No	Plant species/Type of	Name of chemical applied	Effects on essential oil	Effects on other growth	References		
	study		contents/Miscellaneous effects	attributes/Additional properties			
			_ Chlormequat chloride				
			$\begin{array}{c} CI \\ & \\ & \\ & \\ H_{3}C \end{array} \begin{array}{c} CH_{3} \\ CH_{3} \\ CH_{3} \end{array} \begin{array}{c} CI \end{array}$				
1	Baby primrose plant (Primula forbesii)	Chlormequat chloride	Chlormequat chloride possesses growth retarding effects	Reduction in new leaf growth, short height of plants, increased number of floral whorls and significant reduction in length of peduncle	[65]		
2	Mentha arvensis L.	Chloromequat chloride (also named as cycocel)	The damaging effects of cycocel are further ameliorated under the water stressed conditions	Significant improvements in peroxidase activities and percentage of essential oil and menthol contents with slight reduction in biomass yield	[66]		
3	Japanese mint/Wild Mint/Corn mint (<i>Mentha</i> <i>arvensis</i> L. var. Piperascens Mal.)	Chlormequat chloride	The exogenous application of chlormequat chloride can reduce the harmful consequences of acute water shortage in plants	Significant increase in sugar contents, peroxidase activities and menthol contents of essential oil with slight reduction in essential oil percentage, herbage yield and moisture contents	[67]		
			Maleic hydrazide				
	Nalet nydrazide						
1	Cucumber (Cucumis sativus)	Maleic hydrazide	The application of maleic hydrazide reduced the plant height as it acts as a potential herbicide and a known plant growth retardant	Higher chlorophyll contents of plant leaves, increased biomass production, improved sex modifications, higher fruit yield and enhanced fruit settings	[68]		

Table 2: Effects of PGIs on essential oil contents and growth parameters of different plant species

2	Fenugreek (Trigonella	Maleic hydrazide	Specific gene mutation, growth	Significant improvements in seed quality	[69]
	Foenum-Graecum L.)		retarding effects and higher	and morphological characteristics with	
			herbicidal potentials	reduced seed germination and pollen	
				fertility, and lower chlorophyll contents	
3	Chrysanthemum	Maleic hydrazide	The combination of maleic	The application of maleic hydrazide	[70]
	(Dendranthema		hydrazide and potassium showed	maximizes dry and fresh weight of flowers,	
	grandiflorum)		very positive effects on floral	plant height and number of leaves, along	
			quality of plants	with the introduction of early flowering in	
				plants	
			Daminozide		
		H	$\begin{array}{c} O & CH_3 \\ I & I \\ O & I \\ O & H \end{array}$		
1	Ruellia brittoniana	Daminozide	Significant increase in chlorophyll	Daminozide caused significant decrease in	[71]
			contents of leaves and reduction in	plant height, internodal length, leaf area	
			overall plant growth	index, and dry weight of roots, stems and	
				leaves	
2	Fennel (Foeniculum	Alar (also known by different	Exogenous application of alar	Remarkable increase in essential oil contents	[72]
	vulgare)	trade names such as N-	decreased the fenchone percentage	and seed production in fennel	
		dimethyl amino succinamic	and increased the anethole contents		
		acid, SADH, Alar 85, B-995,	in essential oil of plant		
		B-9 and daminozide)			[20]
3	Mentha piperita	Daminozide	Decreased concentration of	Significant increase in essential oil contents	[73]
			menthone and menthol, and	and remarkable improvements in overall	
			increased percentage of	biomass yield	
			isomenthone and neoisomenthol in		
			peppermint		
			Ethephon O H Cl		
1	Winter wheat (Triticum	Ethephon	Ethephon caused appreciable	Improved internodal characteristics and	[74]
	aestivum L.)		increase in plumpness of basal	higher lignin contents	r)
			internodes, wall thickness.		
			breaking strength and internodal		
			diameter		

2	Field mint (Mentha	2-chloroethyl phosphonic	Negative correlation between herb	Significant reduction in biomass yield and	[75]
	arvensis L.)	acid	yield and essential oil contents of	height of plants, indicating the strong growth	
			Mentha arvensis L., with	retarding effects	
			significant improvements in		
			percentage of menthone in		
			essential oil		
3	Mentha piperita	Ethephon	The essential oil of Mentha	Ethephon is a known plant growth retardant	[73]
			piperita showed significant	that reduces essential oil contents and	
			increase in isomenthone and	biomass yield of plants	
			neoisomenthol, and marked		
			decrease in menthone and menthol		
			percentage of essential oil		
			Tetcyclacis		
			Cl		
			(1)		
			N		
			N		
1	Potato (Solanum	Tetcyclacis	The plant growth regulator acts as	The tuber stimulating effects are caused by	[76]
	tuberosum L.)		herbicide and inhibit the growth of	the potential plant growth retardant	[]
			stolon by lowering the level of	r r r r r r r r r r r r r r r r r r r	
			gibberellins		
2	Fenugreek (Trigonella	Tetcvclacis	Tetcyclacis acts as	Better accumulation of cholesterol and 14α -	[77]
_	foenum-graecum L.)		norbornanodiazetine plant growth	methyl sterols contents with reduced growth	[,,]
	Joennin graceum Ell		retardant which decreases the	of seedlings, shoots and roots	
			sanogenin contents in roots of		
			Fenugreek		
3	Moth bean (Vigna	Tetcyclacis	Reduced the high temperature	It makes the plants resistant towards heat	[78]
5	aconitifolia (Jaca)	i ete y enteris	induced lipid oxidation in beans	shock by decreasing the oxidative potential	[,0]
	Marechal cy Iaadia)		and acts as potential herbicide	of ascorbic acid and enhancing the catalytic	
	wareenarev. Jaadla)		and acts as potential herbicide	notentials of catalase and perovidase	
1		1	1	potentials of catalase and perovidase	

Sr.	Plant	Name of	Type of NPs prepared	Method of	Detailed	Characterization	References
No	species/Type of	PGR/Agrochemical		preparation	mechanism/Reaction	techniques	
	study				conditions		
1	Crocantela variety from lettuce seedlings	Indole-3-acetic acid	Chitosan based nanoparticles	Ionic gelation by using sodium tri- poly-phosphate	Chitosin based nanoparticles were prepared by dispersive agitation followed by ultrasonication and centrifugation. The indole-3-acetic acid was loaded by continuous stirring through ultrasonication followed by centrifugation and magnetic agitation for 24 hours	Dynamic light scattering (DLS), polydispersity index (PDI), scanning electron microscopy (SEM), thermogravimetric analysis (TGA) and infrared (IR) spectroscopy	[79]
2	Solanum lycopersicum	Gibberellic acid OH HO HO HO HO HO HO HO OH	Nano alginate/chitosan and nano chitosan/tripolyphosphate containing gibberellic acid	Process of colloidal dispersion	Preparation of nano chitosan/tripolyphosphate containing gibberellic acid using vigorous stirring with the help of peristaltic pump	Dynamic light scattering (DLS), atomic force microscopy (AFM) and nanoparticle tracking analyses (NTA)	[80]
3	Efficiency of removal of plant growth regulator from contaminated aqueous solution	2,4-dichlorophenol	Zinc metal-organic framework having 2- aminoterephthalic acid and 1,4-bis(4-pyridyl)- 2,3-diaza-1,3-butadiene	The process of sonication facilitated by ultrasound	Preparation of single crystal zinc metal-organic framework having 2- aminoterephthalic acid and 1,4-bis(4-pyridyl)- 2,3-diaza-1,3-butadiene followed by controlled crystal growth using sonochemical process	Thermogravimetric analysis (TGA), Brunauer–Emmett– Teller (BET), Infrared spectroscopy, X-ray diffraction (XRD) and Scanning electron microscopy (SEM)	[81]
4	Seeds of Phaseolus vulgaris L.	Gibberellic acid	Alginate/chitosan and chitosan/tripolyphosphate based nano-formulations	The process of gelation and ionotropic pre- gelation	Preparation of alginate/chitosan nanoparticles using ionotropic pre-gelation and encapsulation of gibberellic acid using chitosan/tripolyphosphate	Zeta potential and poly dispersity index (PDI)	[19]

Table 3: Nanostructured PGRs, their methods of preparation and reaction conditions

		HO _{H3C} OO			based nano-formulations aided by gelation method		
5	Controlled release of plant growth regulators	Phenylacetic acid	Preparation of polysaccharide beads from xylan, cellulose nanocrystals, starch, cellulose powder and beads of alginate reinforced with kaolin clay modified with polyethylenimine	Slow entrapment of phenylacetic acid in complex network of cellulose	Preparation of polysaccharide beads by homogeneous suspension using crosslinking agent, followed by drying (48 hours) and soaking in solution of phenylacetic acid (48 hours)	Energy dispersive spectroscopy (EDS), Ultraviolet-visible (UV–Vis) spectroscopy, Scanning electron microscopy (SEM) and Tensile testing (TT)	[82]
6	Wheat seeds	1-Naphthylacetic acid	Conjugated nanospheres of 1-naphthylacetic acid and silica	The covalent crosslinking reactions followed by hydrolyzation and poly- condensation process	Conjugation of 3- aminopropyltriethoxysilan e and 1-naphthylacetic acid through covalent cross-linking and hydrolyzation in the presence of tetraethyl orthosilicate to form 1- naphthylacetic acid and silica nano-spheres	High performance liquid chromatography (HPLC), Scanning electron microscopy (SEM), Fourier transform infrared (FTIR) spectroscopy and UV–Vis spectroscopy	[83]
7	Preparation of nano- formulations of herbicide	$\begin{matrix} \text{Imazaquin} \\ & \overset{\text{OH}}{\longleftarrow} \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & & \\ & & & \\ & & & & \\$	Encapsulation of imazaquin using starch and chitosin beads reinforced with alginate	The method of external gelation	Starch/alginate beads and chitosan/alginate beads were prepared by extrusion of beads in calcium chloride followed by starch gelatination at 75°C	Fourier transform infrared (FTIR) spectroscopy, Scanning electron microscopy (SEM) and Differential scanning calorimetry (DSC)	[84]

8	Preparation of herbicides based nanoformulation s	Atrazine CI H_3C N N N CH_3 H H H	The microspheres of poly(hydroxy-butyrate- co-hydroxy-valerate) loaded with atrazine	Microspheres were prepared by solvent evaporation method and emulsification process	Solubilization of atrazine with poly(hydroxy- butyrate-co-hydroxy- valerate) in the presence of polyvinyl alcohol followed by heating, evaporation and centrifugation of nano- suspensions	Ultraviolet (UV) spectrophotometry and Scanning electron microscopy (SEM)	[85]
9	Water soluble pesticides showing controlled release and slow delivery of bioactive compound	Validamycin HO HO HO HO HO HO HO HO HO HO HO HO HO	Silica based porous hollow nanoparticles loaded with validamycin	Loading of agrochemical using supercritical fluid method	Use of sol-gel method for the preparation of hollow silica nanoparticles by heating upto 353 K followed by adsorption of validamycin at 40 MPa for 5 hours	High performance liquid chromatography and ultra-violet spectrophotometry, Transmission electron microscope (TEM), Thermo-gravimetry- differential thermal analysis/differential scanning calorimetry (TG-DTA/DSC), Infrared (IR) spectroscopy and X-ray photoelectron spectroscopy (XPS)	[86]

Sr. No	Name of agrochemical and	Molecular structure	Photosensitizers	Techniques/Apparatus	Experiments conducted	References
1	Maleic hydrazide (Herbicide)	O NH NH	Humic acid and vitamin B ₂ /Riboflavin	The steady state fluorescence by using Spex Fluoromax spectrofluorimeter and absorption measurements using Hewlett Packard 8452A diode array spectrophotometer	Time resolved phosphorescence detection, continuous photolysis and laser flash photolysis	[87]
2	2-(1-naphthyl) acetamide (Plant growth regulator)	NH2	Rose Bengal	Liquid chromatography-mass spectrometry-mass spectrometry (LC/MS/MS), High performance liquid chromatography-mass spectrometry-mass spectrometry (HPLC/MS/MS), Ultraviolet– Visible spectroscopy and Liquid chromatography-mass spectrometry (LC/MS)	Assessment of hydrogen peroxide free radical formation and nanosecond laser flash photolysis	[88]
3	Indole-3-carboxylic acid (Plant growth regulator)	HO N H	The reference sample contains haricot bean	Ultraviolet–visible spectroscopy using Beckman instrument (model: 3600) and photochemical reactor containing an oriel light source equipped with (i) tungsten-lamp and (ii) oriel power supply box, with a magnetic agitator and (iii) a quartz cell placed at 60 cm from the lamp	Determination of half- life, rate constants, caulogenesis, rhizogenesis and photodegradation kinetics	[89]

Table 4: Photodegradation studies of PGRs and other agrochemicals

Sr.	Name of	Molecular structure	Type of nanoparticle	Technique of	Instrumental specifications and	References
No	agrochemical			thermal	processing conditions	
	and category			analysis used		
1	Azoxystrobin		Metal organic framework	Thermal	Thermal gravimetric analysis was	[90]
	(Fungicide)		containing porous iron based	gravimetric	performed by PerkinElmer Pyris	
			nano-structures	analysis	Diamond having 20 mL/min and 30	
		N N O		(TGA)	mL/min flow rate of nitrogen for	
		CH ₃			sample and reference. All readings	
					were taken using nitrogen steam	
					having a heating rate of 10°C/min to	
					550°C at the end	
2	Imazethapyr	CH ₃	Biopolymeric beads containing	Thermal	Thermal gravimetric analysis was	[91]
	(Herbicide)		alginate cellulose mixture	gravimetric	performed by Shimadzu Analyzer	
				analysis	(model: DTG60, Japan) with	
				(TGA)	scanning rate: 10°C/min in	
		Ö HN N			nitrogenous atmosphere and flow	
					rate: 20 mL/min with a temperature	
					ranging from 25 to 600°C	
2	Cibbonallia a sid	H ₃ C ⁻ CH ₃	Chita any trin alamba and ata and	Differential	Chita and Aring hards and at and	[10]
3	(Dippereinic acid	OH OH	chitosan/tripolyphosphate and	Differential	chitosan/impolyphosphate and	[19]
	(Flant glowin		boads	colorimetry	were characterized by differential	
	regulator)		Utaus	(DSC)	scanning colorimetry using	
		HO _{H3C}		(DSC)	aluminum cups with a constant flow	
		0			rate of nitrogen (50 mL/min) and	
					heating from 20 to 350°C	

Table 5: Thermaldegradation studies of PGRs and other agrochemicals

Sr.	Name of	Molecular structure	Type of nanoparticle	Encapsulation	Technique used and	References
No	agrochemical and			efficiency	instrumental specifications	
	category					
1	Gibberellic acid	ОН //	Nano alginate-chitosan and	Entrapment	Ultrafiltration membranes were	[80]
	(Plant growth	ОЩ	nano chitosan-	efficiency of nano	used with the dense cellulosic	
	promoter)		tripolyphosphate having	chitosan-	network with the approximate	
			gibberellic acid	tripolyphosphate is	pore size of 30 kDa followed by	
		HO _{H₃C}		90% and nano	quantitative analysis of entrapped	
				alginate-chitosan is	gibberellic acid through high	
				100%	performance liquid	
					chromatography	
2	Diammonium		The superabsorbent	Almost 80% release	The standard curve of pure	[92]
	hydrogen		composites containing starch	of diammonium	diammonium hydrogen phosphate	
	phosphate	O=P-OH	and alginate	hydrogen phosphate	drawn by using high performance	
	(Phosphorous			from dense network	liquid chromatography at 20:80	
	fertilizer)			of superabsorbent	volume/volume methanol to water	
				composites after 30	ratio. Separation was achieved by	
				days of treatment	a thin column at the flow rate of 1	
					mL/min at 30°C followed by the	
					ultraviolet-visible (UV-Vis)	
					spectrophotometric detection at	
					254 nm	
3	Gibberellic acid	ОН //	Alginate-chitosan and	The encapsulation	Analysis of prepared solution	[19]
	(Plant growth	ОН	chitosan-tripolyphosphate	efficiency of 50	using ultrafiltration device having	
	regulator)			µg/mL solution was	cellulosic membrane (pore size:	
				found to be 100%	30 kDa) followed by	
		HOHAC			centrifugation for 30 minutes. The	
		3-			filtrate was analyzed by high	
		U U			performance liquid	
					chromatography for the	
					quantification of residual	
					hormone, not associated with the	
					NPs	

Table 6: Leaching potentials of PGRs and other agrochemicals

Sr.	Plant material or type	Type of cell line used	Cytotoxic assay and detailed	Probable outcomes of	References
No	of NPs		process	research	
1	Mentha arvensis L.	Hepatocellular carcinoma in	The MTT cytotoxic assay was	The results obtained	[93]
	(A rich source of	HepG ₂ cells of human beings	performed by using HepG ₂ cell	confirmed the anticancer	
	rosmarinic acid fraction)		line by treating with rosmarinic	potential of rosmarinic acid	
			acid rich fraction of Mentha	by G _o /G ₁ phase arrest and	
			arvensis L. The entire setup	synchronizing the ERK2, Bax	
			was kept in the incubator for 24	and Bcl-2 mRNA	
			hours	expressions, resulting in	
				apoptosis of severe	
				hepatocellular carcinoma	
2	Mentha spicata essential	Breast adeno-carcinoma MCF-7	The MTT cytotoxic assay was	The obtained results showed	[94-95]
	oil	cell line, kidney carcinoma	performed by dissolution of	that the essential oil of	
		A498 cell line, ductal breast	dimethyl sulfoxide in essential	Mentha spicata possesses	
		epithelial tumor T47D cell line,	oil along with the analyte of	enough potential to neutralize	
		renal cell carcinoma Caki cell	interest, followed by the	the free radical species of	
		line, EBV-negative Burkitt's	incubation at 37°C for the	cancerous cells, thereby	
		lymphoma BJAB cell line, ,	period of 48 hours	acting as an anti-cancer agent	
		prostate adenocarcinoma PC-3		through reversal of	
		cell line, colon adeno-		tumorigenesis. These	
		carcinoma Caco-2 and		processes helps to avoid the	
		epithelial carcinoma HeLa cell		deleterious effects of	
		line		chemotherapy	
3	Synthesis of green silver	MDA-MB-231 and MCF7 cells	The survival rate of treated and	The results obtained by the	[96]
	nanoparticles using		untreated cells were studied by	MTT cytotoxic assay showed	
	Mentha arvensis L.		MTT cytotoxic assay by the	that the nanoparticles	
			use of fluorescence activated	exhibited far better cytotoxic	
			cell sorting through	effects against breast cancer	
			fluorescence microscopy	with increase in the	
				population of G_1 cells.	
				Hence, these nanoparticles	
				can preferably be used as	
				they are non-mutagenic and	
				least toxic in nature	

Table 7: Cytotoxic effects of different Mentha species

Sr.	Name of plant/Type of	Name of PGR/Type of	Instrumental specifications	Probable outcomes/Obtained results	References
No	sample/Purpose of analysis	Agrochemical			
		Fourie	r transform infrared (FTIR) spectro	oscopy	
1	Cherry tomatoes	Glyoxylic acid monohydrate, 1-naphthaleneacetic acid,	IR spectra were recorded by using an ona Vertex70 FTIR	Among all the peaks obtained in FTIR spectra, the stronger peaks at 3384 cm ⁻¹ represents –OH group,	[97]
		indole-3-butyric acid and	spectrometer (provided by Bruker,	1816 cm ⁻¹ and 1630 cm ⁻¹ represents C=C stretching	
		indole-3-propionic acid	Karlsruhe, Germany) in the range	vibrations, 1736 cm ⁻¹ represents C=O and 1069 cm ⁻¹	
			of 500 cm ⁻¹ to 4000 cm ⁻¹	represents the C-O-C stretching vibrations	
2	Plant based tissues, fruits	2,4-dichlorophenoxyacetic	FTIR spectra were obtained by	The spectral peaks at 3596 cm^{-1} and 2927 cm^{-1}	[98]
	and vegetables	acid, 1-naphthaleneacetic	using the FTIR spectroscopy	indicates the presence of methacrylic acid and β -	
		acid, indole-3-butyric acid	through NICOLET IS10	cyclodextrin, while the peaks at 1731 cm ⁻¹ represents	
		and indole-3-acetic acid	instrument	C=O stretching vibration, 1636 cm ⁻¹ represents O-H	
				deformation, 1149 cm^{-1} represents O–C(O)–C bonds	
				and the peaks at 2989 cm^{-1} , 2957 cm^{-1} , 1458 cm^{-1} , and	
				1387 cm ⁻¹ indicates the other stretching vibrations	
3	Solid phase magnetic	Tebuconazole, uniconazole,	The FTIR spectrums were obtained	The characteristic peak at 578 cm ⁻¹ represents Fe–O–	[99]
	adsorbents based on β -	paclobutrazol, atrazine, 2,4-	through a Tensor 37 (Bruker	Fe, 1082 cm^{-1} and 3420 cm^{-1} indicates Si–O–Si and	
	cyclodextrin	dichlorophenoxyacetic acid,	spectrometer company, Germany)	O–H stretching vibrations, 1628 cm ⁻¹ represents	
		1-naphthaleneacetic acid,		bending vibration of N–H, 3500–3300 cm ⁻¹ shows	
		simazine, 4-		stretching vibration of N–H, and 1716 cm ⁻¹ and 1388	
		chlorophenoxyacetic acid,		cm^{-1} represents the stretching vibration of C=O and	
		and 4-fluoro-phenoxyacetic		bending vibration of O–H groups	
		acid			
		Laser	induced breakdown spectroscopy ()	LIBS)	
1	Seaweed	Fertillizers based on seaweed	LIBS spectra was obtained by a	Stronger emission lines were obtained for strontium,	[45]
			Nd:YAG laser (wavelength 1064	sodium, potassium, calcium, magnessium, and	
			nm, pulse duration 6ns, Litron	manganese, and some trace elements like silicon (at	
			Lasers, Rugby, United Kingdom)	288.15 nm and 390.53 nm) and iron (at 382.04 nm,	
				3/5.82 nm, 3/4.95 nm, 3/4.83 nm, 3/4.56 nm, 3/3./1	
-				nm, 3/3.49 nm and 3/1.99 nm)	F1003
2	Quick method of	Fertillizers	The LIBS spectra was generated	The strong characteristic lines of the elemental	[100]
	identification of potassium,		by a Q-switched Nd: YAG pulsed	phosphorus were found to be at 255.5 nm, 255.3 nm,	
	nitrogen and phosphorous		laser with the following	253.6 nm, 253.4 nm, 215.4 nm, 214.9 nm and 213.5	
			specifications (ICE450, 1064 nm,	nm, elemental potassium at 769.9 nm, 766.5 nm and	
			o ns puise duration, Big Sky Laser	404.7 nm, and elemental nitrogen at $8/1.8$ nm, $8/1.2$	
			LICA suith an alternative number of	nm, 8/0.5 nm, 862.9 nm, 859.4 nm, 856./ nm, /46.8	
			USA with an alternative name of	nm, 744.2 nm and 742.4 nm	
			Quantel Laser		

Table 8: Characterization techniques for nanostructured PGRs, nanofertilizers and other agrochemicals

3	Estimation of potassium in fixed concentration nutrient media	Fertillizers based on biochar	The calcium and potassium were analyzed by Q-switched Nd:YAG laser, capable of emitting at 1064 nm (Quantel, Big Sky Laser, Ultra 50) and operating at maximum energy of 50 mJ with a repetition rate up to 20 Hz and a pulse duration of 20 ns, thereby arracting the laser ener diameter of	The LIBS spectra showed the strong peaks for calcium- I at 431.86 nm, 430.25 nm, 428.94 nm and 428.30 nm, and for potassium-I at 404.64 nm	[101]	
			about 0.5 mm			
		Α	tomic absorption spectrometry (AA	S)		
1	Trace metal determination in soil	Complete environmental analysis of contaminated soil samples	For the elemental analysis, a flame atomic absorption spectrometer (model: AAnalyst 700, PERKIN ELMER-SCIEX) was used in acetylene/nitrous oxide flame and air/acetylene flame for determination of aluminium and cadmium, cupper, nickle and zinc	The traces of metallic elements in soil samples were successfully extracted by the "ultrasound-assisted- extraction" method, using hydrochloric acid, hydrofluoric acid and nitric acid. Among all the acids, hydrofluoric acid proved to be the most efficient solvent for the extraction of zinc, nickle, cupper, cadmium and aluminium to be analyzed by the atomic absorption spectrometry	[102]	
2	Qualitative and quantitative determination of cadmium in oyster and water samples	Solid-phase and magnetic micro-extraction technique by using CoFe ₂ O ₄ nanoparticles	For the analysis of cadmium in given sample, a Perkin Elmer Instruments (Norwalk, CT, USA) with the model AAnalyst 200 flame atomic absorption spectrometer was used	As a result of complete elemental analysis by atomic absorption spectrometry, it was confirmed that the amount of cadmium in the experimental water samples and oyster samples was in the range of 2.75 μ g/L to 4.8 μ g/L and 4.6 mg/kg to 7.8 mg/kg, respectively	[103]	
3	Trace metal analysis in organic fertillizers	Assessment of zinc, lead, nickle, manganese, copper, chromium, cobalt and cadmium in organic fertilizers	For the determination of trace metals, a Varian Spectra 220 (model: Mulgrave, Australia) flame atomic absorption spectrometer was used in a sequential mode, equipped with the deuterium background corrector and multi-element hollow cathode lamp	During this experiment, nitric acid, hydrofluoric acid and hydrochloric acid were used for the solvent extraction of trace metals from analytical samples. Among all these variables, hydrofluoric acid proved to be best for the extraction of traces of metals due to the high dissolution potential. Thus extracted metals were analyzed by the atomic absorption spectrometry	[104]	
Scanning electron microscopy (SEM)						
1	Synthesis of novel PGRs based on the potential derivatives of dehydroamino acids followed by the	1-methyl-3-methylamino- maleimide, Z-isomer of potassium salt of 2-amino-3- methoxycarbonylacrylic acid	The surface morphology and particle size of the nano-structured PGRs were studied by scanning electron microscopy (FE-SEM,	The results obtained by the SEM analysis showed that the wet encapsulations of PGRs do not have significant effects on the overall particle size. The size of wet beads ranged from 1706.07 µm to 2120.64 µm and dry	[105]	

	encapsulation inside the	and 2,3-dehydroaspartic acid	model: JSM-7000F, Jeol Ltd.,	beads ranged from 745 µm to 810 µm. The surface of	
	biopolymeric beads	dimethyl ester	Japan)	dried nanoparticles was quite rough and wrinkled	
2	Fabrication of biochar based	2,4-dichlorophenoxyacetic	Some morphological	As per the data obtained by SEM analysis, the	[106]
	nanocarriers for sustained	acid	characteristics and elemental	hydrodynamic particle size of the nano-encapsulated	
	release of nano-herbicides		analysis of the encapsulated nano-	herbicide was around 256.5 nm which indicates the	
			herbicide was done by a VEGA 3	high loading capacity, enhanced herbicidal efficiency,	
			TESCAN, scanning electron	better penetration potential, controlled release and	
			microscope (SEM), coupled with	sustained delivery of agrochemical, leading towards the	
			EDAX for the dual functions	sustainable developments	
3	Synthesis of nano-structured	Tricyclazole	To study the topological	As per the conventional approximations, smaller	[107]
	tricyclazole	-	properties, to determine the	particle size of agrochemicals enhances the rate of	
	(Fungicide)		particle size and to assess the	dissolution, lower the leaching of chemicals, enhances	
			morphological characteristics of	the surface area and improves the penetration potential.	
			nano-fungicide, scanning electron	Highly magnified images of SEM indicated the	
			microscopy (model: Zeiss Evol8)	different shapes of tricyclazole leading to the	
			was used with the variable	agglomeration of particles and formation of nano-	
			magnification ranges using the	clusters. The particle size distribution of nano-	
			sputtering coater and electron	tricyclazole is found to be in the range of 60 nm to 95	
			beam released by electron gun	nm	
			Thermal analysis (TA)		
1	Preparation of encapsulated	Spinosad	Differential scanning calorimeter	The prepared chitosin based encapsulating structure of	[108]
	chitosin based pesticidal	(Pesticide)	(model: Shimadzu, DSC-60 Plus,	nanoparticles showed the initial volatilization of	
	formulations for slow		Japan) was used having a	potentially adsorbed water at the temperature range of	
	release of spinosad by		temperature range of 30°C to	50°C to 230°C, rapid weight loss at temperature range	
	coprecipitation		400°C with 10°C/min increase in	of 230°C to 312°C and relatively slow weight loss at	
			temperature, and TGA curves were	temperature range of 312°C to 430°C. Similarly,	
			obtained by TGA/DSC	spinosad loaded encapsulated nanoparticles showed the	
			simultaneous thermal analyzer	rapid weight loss at the temperature range of 230°C to	
			(model: Discovery, SDT 650,	312°C, indicating the significant improvements in	
			USA) with temperature range of	thermal stability of prepared nano-formulations. The	
			30°C to 600°C at the rate of	decomposition temperature is around 312°C to 600°C	
			10°C/min increase in temperature		
2	Synthesis of ecofriendly	Biochar	Thermal analysis was performed	In the TGA curves, first peak at 147.52°C showed the	[109]
	corncob biochar based	(Nano-fertilizer)	by using the TGA/DSC (Model:	rapid weight loss due to higher moisture contents,	_
	nano-composites to ensure		SDT-Q600, Germany) in order to	second peak at 335.13°C indicated gradual weight loss	
	the sustained delivery of		check the thermal stability of	and third peak at 481.25°C represented pyrolytic	
	agricultural chemicals		prepared nano-formulations	degradation of sample with the release of methane,	
				carbon monoxide and carbon dioxide	
3	Preparation of a novel	Brassinosteroid	DSC was performed by a	The results obtained by the DSC/TGA curves showed	[110]
	brassinosteroid modified	(Growth regulating hormone)	PerkinElmer Differential Scanning	that major endothermic processes mainly relevant to	_

	polyethylene glycol micelles		Calorimeter Pyris 1 (PerkinElmer,	the melting of prepared nano-formulations are in the	
	for targeted delivery of		Inc., Waltham, MA, USA) and	range of 38°C to 64°C while exothermic processes	
	agrochemical		TGA curves were obtained by	including the pyrolytic degradations are in the range of	
	_		Netzsch TG 209 C Iris system	225°C to 300°C with weight loss up to 98%	
			X-ray diffraction (XRD)		
1	Synthesis of interpenetrating	Hydrogels integrated with	For the XRD analysis of prepared	In the XRD pattern, several sharp lines were obtained	[111]
	dense polymeric networks	natural soil colloid – a	hydrogels and colloidal soil	indicating the multiple mineral elements of the soil	
	of hydrogels potentially	potential carrier for slow	particles, the samples were	colloides. Some phase peaks of pure hydrogels in XRD	
	integrating natural soil	release of agrochemicals	analyzed at diffraction angles	pattern were relatively broad, indicating the poor	
	colloids via covalent cross-	-	ranging from 2° to 70° with the	crystallinity of hydrogel matrix. In case of hybrid	
	linking of polyacrylamide		help of x-ray diffractometer	hydrogels, 7.5 wt% of the soil colloids were identified,	
	and ionic crosslinking of		(model: D8 Advance, Bruker) with	indicating the presence of soil colloids inside the dense	
	alginate		CuK α radiation (at λ =0.1548 nm)	polymeric network	
	_		at 40 kV and 40 mA		
2	Preparation of	Permethrin and spinosad –	For the powdered XRD analysis of	The XRD pattern of nano-formulations exhibited a	[112]
	biocompatible chitosin	two potential pesticides for	nano-structured pesticides, model:	strong characteristic peaks at 20° indicating the purely	
	based nanoparticles loaded	proper insect pest	Panalytical's X'Pert Pro was used	crystalline nature of chitosan. On the other hand, the	
	with spinosad and	management	by taking 2 g of finely powdered	broad peaks of loaded chitosan nanoparticles indicate	
	permethrin		sample on a glass slide having	the amorphous nature and incoherent scattering of x-	
			uniform layer of nanoparticles	rays. The results showed that crosslinking destroyed	
				the crystalline structure of loaded chitosin	
				nanoparticles	
3	Preparation of	Nitrogen-phosphorous-	The XRD patterns of cellulosic	The peak at 22.5° confirmed silica, peaks at 37°, 36°,	[113]
	carboxymethyl cellulose	potassium (NPK)	nanoparticles were obtained by	32°, 29°, 25° and 23° confirmed urea, peaks at 49°,	
	based nanocomposites for	(Fertilizer)	using a Siemens D500 x-ray	47°, 46°, 34°, 31°, 24° and 17.5° confirmed potassium	
	slow release of basic		diffractometer (model: Siemens	dihydrogen phosphate and peaks at 69°, 62°, 60.5°,	
	nutrient elements by		AG, Karlsruhe, Germany) with a	58.5°, 51.5°, 45°, 38.5°, 34°, 29°, 23° and 17.5°	
	fertillizers and water		CuK α radiation in a 2 θ range of 2°	confirmed ammonium dihydrogen phosphate in the	
	retention in soil		to 70° angle	given samples	

7.4. Scanning electron microscopy (SEM)

Scanning electron microscopy (SEM) has widely been used to determine the shape and surface morphology of nanoparticles (NPs) in various disciplines all over the world. SEM is regarded as the most effective method for the analysis of organic and inorganic samples, having particle size in the micrometer (μ m) and nanometer (nm) range. The extent of magnification of some modern SEM instruments can reach up to 300,000x and even 1000000 in producing the images for a wide range of samples. SEM works with energy dispersive x-ray spectroscopy (EDS) for qualitative and semiquantitative results. Together, both these techniques can produce the basic information about chemical composition of scanned samples [47].

7.5. Thermal gravimetric analysis (TGA)

Thermal gravimetric analysis (TGA) is one of the most important types of thermal analysis (TA), capable of providing the valuable informations regarding the change in mass and chemical composition of prepared nanoparticles (NPs). In this method, nanomaterials are heated up to degradation temperature and their components are decomposed and vaporized with the passage of time. The loss of mass with the change in temperature is recorded with the help of TGA device, and quantity, type and mass of prepared nanoparticles (NPs) are determined by comparing it with the starting material. In the case of "micro-thermogravimetric analysis" (μ -TGA), the amount of sample generally varies in between one nano-gram to one micro-gram, thereby improving the detection limit of conventional TGA up to great extent [43].

7.6. X-ray diffraction (XRD)

In the x-ray diffraction (XRD), scattering of x-rays is used to elucidate the crystallinity of crystalline and semicrystalline materials. The scattering of x-rays by periodic arrangement of atoms, give rise to the definite diffraction patterns, thereby providing the qualitative imaging of atomic arrangements inside the crystal lattice. In case of powder xray diffraction (P-XRD), both the precursors and end products can simultaneously be characterized with complete qualitative representation of micro-structural behaviours. This method is quite convenient as compared to the single crystal XRD that requires individual crystal for analysis. In addition, XRD is the non-destructive method for crystallographic determination of a crystal [48].

The use of cost effective and environmentally benign plant growth regulators (PGRs) is the most preferred option in modern agriculture. Plant growth regulators (PGRs) not only enhance the biomass yield, essential oil percentage and amount of major secondary metabolites, but also improve stress responses and environmental resistance in plants. However, long duration of sunlight, high temperature and cytotoxic nature of plant growth regulators (PGRs) can cause severe damage to plants and environment. The excessive leaching of agrochemicals not only pollutes the natural water reservoirs, but also damages the life quality of aquatic animals. Therefore, use of nanotechnology holds great potential in this regard. Nano encapsulated and polymeric plant growth regulators (PGRs) ensure the controlled release, long persistence and targeted delivery of applied chemicals to plants. Hence, nanotechnology can work wonder to increase the production of food crops and essential oil contents of aromatic medicinal plants.

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