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# Efficient and reusable heterogeneous nano sized CeO<sub>2</sub> catalyst for the synthesis of 1, 8-dioxo-octahydroxanthenes

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### Abstract

Key words: 1,8-Dioxo-octahydroxanthene, 5,5-Dimethyl-1,3-cyclohexanedione, Solvent free, Nano sized Cerium oxide

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### 1. Introduction

Secondary metabolites are important natural products with many applications in medicine and technology [1]. Among the natural heterocyclic compounds, xanthene derivatives have an extensive range of biological and pharmaceutical applications as antiviral, antibacterial and anti-inflammatory activities [2-4]. Also, their importance as dyes in laser technology, pH-sensitive fluorescent and versatile synthons caused to obtain a considerable interest in researches [5-6]. The synthesis of 1,8-dioxooctahydroxoxanthene derivatives with the same reagents using InCl<sub>3</sub>.4H<sub>2</sub>O in ionic liquids, solid state condensation, aqueous media containing NH<sub>2</sub>SO<sub>3</sub> and micro wave media [7-9] were studied but these methodologies have some defects like low yield, prolonged reaction time and using of toxic organic solvents. Recently some research groups reported some efficient methodologies using catalysts such as ferric-montmorilonite [10], nano sized TiO<sub>2</sub> [11], sodium hydrogen sulfate supported on silica chloride [12], magnetite-silver nanoparticles [13-14], amberlyst-15 [15], polyaniline-p-toluenesulfonate salt [16] and TMSCI [17] for removing or lowering disadvantages in synthesis of xanthenedione derivatives.

Nowadays nano catalysts showed very valuable role in organic synthesis because of obtaining high yield, simple preparation and isolation, low expense, high recovery, reusability and green properties, therefore, a lot of interest in various chemical transformations using catalysts under heterogeneous conditions has been increased [18-23]. For unusual physico chemical properties of nano sized metal oxides, they can exist with numerous surface sites and enhanced surface reactivity such as crystal corners, edges or ion vacancies [24-26]. The small size and large specific surface area of nano sized  $CeO_2$  allow for certain unique and unusual physico chemical properties [27-28]. Cerium oxide is a prominent material for various kinds of industrial applications related to catalysis [29-30].

Through our further studies on the heterocycles [2, 31-33], we investigated the effect of catalytic activity of nano sized  $CeO_2$  for preparation of 1,8-dioxo-octahydroxanthenes using various aromatic aldehydes (Figure. 1).

# 2. Material and Methods

Nano powder  $CeO_2$  with particle size of 25 nm was purchased of sigma Aldrich company. <sup>13</sup>C NMR and <sup>1</sup>H NMR spectra were recorded on Brucker, 100 and 250 MHz using TMS as an internal standard. Chemical shifts are reported in ppm, and coupling constants are reported in Hz. IR spectra were recorded on a Shimadzu 470 spectrophotometer. TLC was performed on Merckprecoated silica gel 60-F254 plates.

# 2.1. General Procedure for the synthesis of 1,8-dioxododecahydroxanthenes

The synthesis of 1,8-dioxo-dodecahydroxanthenes was according to method reported in literature with some modifications [2]. A mixture of 5,5-dimethyl-1,3-cyclohexanedione (2 mmol), aldehyde (1 mmol) and Nano CeO<sub>2</sub> (0.15 g) was heated at 110 °C for 12 h. After cooling the reaction mixture, it washed with CHCl<sub>3</sub> (10 ml). The solvent was evaporated and the crude product was recrystallized from EtOH to obtain the pure product. All products were known and characterized by comparison of their physical data with those reported in literatures [2, 13]. The spectral data of some representative 1,8-dioxo-dodecahydroxanthenes are given:

#### *3,3,6,6-Tetramethyl-9-(4-N,N-dimethylphenyl)-1,8dioxooctahydroxanthene (3f)*

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  1.02 (6H, s), 1.11 (6H, s), 2.15-2.28 (4H, J = 16.3 Hz, q), 2.47 (4H, s), 2.89 (6H, s), 4.68 (1H, s), 6.64-7.15 (4H, m).

#### *3,3,6,6-Tetramethyl-9-(3,4-dimetoxyphenyl)-1,8dioxooctahydroxanthene (3i)*

IR (KBr): 3008, 2959, 1665, 1464, 1361, 1199, 1167, 856 cm<sup>-1</sup>. <sup>1</sup>H NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  0.93 (6H, s), 1.03 (6H, s), 2.13 (4H, s), 2.39 (4H, s), 3.71 (3H, s), 3.77 (3H, s), 4.62 (1H, s), 6.66-6.82 (3H, m). Anal. Calcd for C<sub>25</sub>H<sub>29</sub>O<sub>5</sub>: C, 73.33; H, 7.14. Found: C, 73.40; H, 7.18.

### *3,3,6,6-Tetramethyl-9-(2-chloro-6-flourophenyl)-1,8dioxooctahydroxanthene (3j)*

IR (KBr): 3070, 2957, 1703, 1653, 1452, 1376, 1240, 1186, 787 cm<sup>-1</sup>. <sup>1</sup>H NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  0.92 (6H, s), 1.04 (6H, s), 2.26 (4H, s), 2.46 (4H, s), 4.85 (1H, s), 6.54-7.19 (3H, m). Anal. Calcd for C<sub>23</sub>H<sub>23</sub>O<sub>3</sub>ClF: C, 68.74; H, 5.77. Found: C, 68.80; H, 5.80.

#### 3. Results and Discussion

Firstly, the optimized amount of nano catalyst was obtained as 0.15 g through the reaction between reactants. To study the effect of catalyst, the reaction of 5,5-dimethyl-1,3-cyclohexanedion (2 mmol) and benzaldehyde (1 mmol) and nano sized  $CeO_2$  (0.15 g) was selected as model reaction under thermal and solvent free conditions, (Figure 1). Also, the aromatic aldehydes containing electron realizing and withdrawing groups were employed and yield was obtained, (Table 1). In all cases the reaction gives the products in good yields and prevents problems associated with solvent utilizing such as toxicity, cost, handling, safety and pollution. Nano Cerium oxide works under heterogeneous conditions. It can easily be handled and removed from the reaction mixture by simple filtration. The recovered catalyst was reused consecutive five times with insignificant loss and a minimum variation in yields of the product (Figure 2).



Figure 1. 1,8-dioxo-octahydroxanthenes from different aromatic aldehydes using nano sized CeO<sub>2</sub> as a heterogeneous catalyst



Figure 2. Reusability of nano sized  $CeO_2$  as a heterogeneous catalyst for synthesis of 1,8-dioxo-octahydroxanthenes.

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Entry	Ar	Product	Yield [%] <sup>a</sup>	Mp (°C)
1	$C_6H_5$	<b>3</b> a	78	203-205
2	$2-ClC_6H_4$	3b	92	225-227
3	$2,4-(Cl)_2C_6H_3$	3c	88	253-256
4	$3-NO_2C_6H_4$	3d	71	169-171
5	$3-BrC_6H_4$	3e	74	190-192
6	$4-N(Me)_2C_6H_5$	3f	83	221-224
7	$2\text{-OCH}_3C_6H_4$	3g	80	183-185
8	$4-OCH_3C_6H_4$	3h	83	240-242
9	$3,4-(OCH_3)_2C_6H_3$	3i	87	182-184
10	2-Cl-6-F-C <sub>6</sub> H <sub>3</sub>	<u>3</u> j	90	117-120

**Table 1:** Synthesis of 9-aryl substituted 1,8-dioxo-octahydroxanthenes using nano CeO<sub>2</sub>

<sup>a</sup>Yield refer to the pure isolated products.

#### 4. Conclusion

In conclusion, we described the application of  $CeO_2$  nano sized particles as a reusable and efficient catalyst for the synthesis of 1,8-dioxo-octahydroxanthenes. These compounds were prepared through treatment of dimedone with various aromatic aldehydes under solvent-free and thermal conditions. The simplicity together with the use of inexpensive, non-toxic and environmentally benign catalyst are remarkable features of the procedure.

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